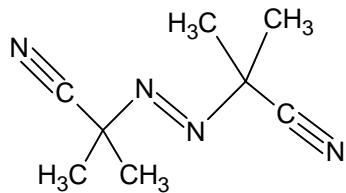


産業技術総合研究所 高エネルギー物質研究グループ
発熱分解エネルギー測定の標準化 熱分析結果

Azobis butylonitrile

(CH₃)₂C(CN)N=NC(CN)(CH₃)₂

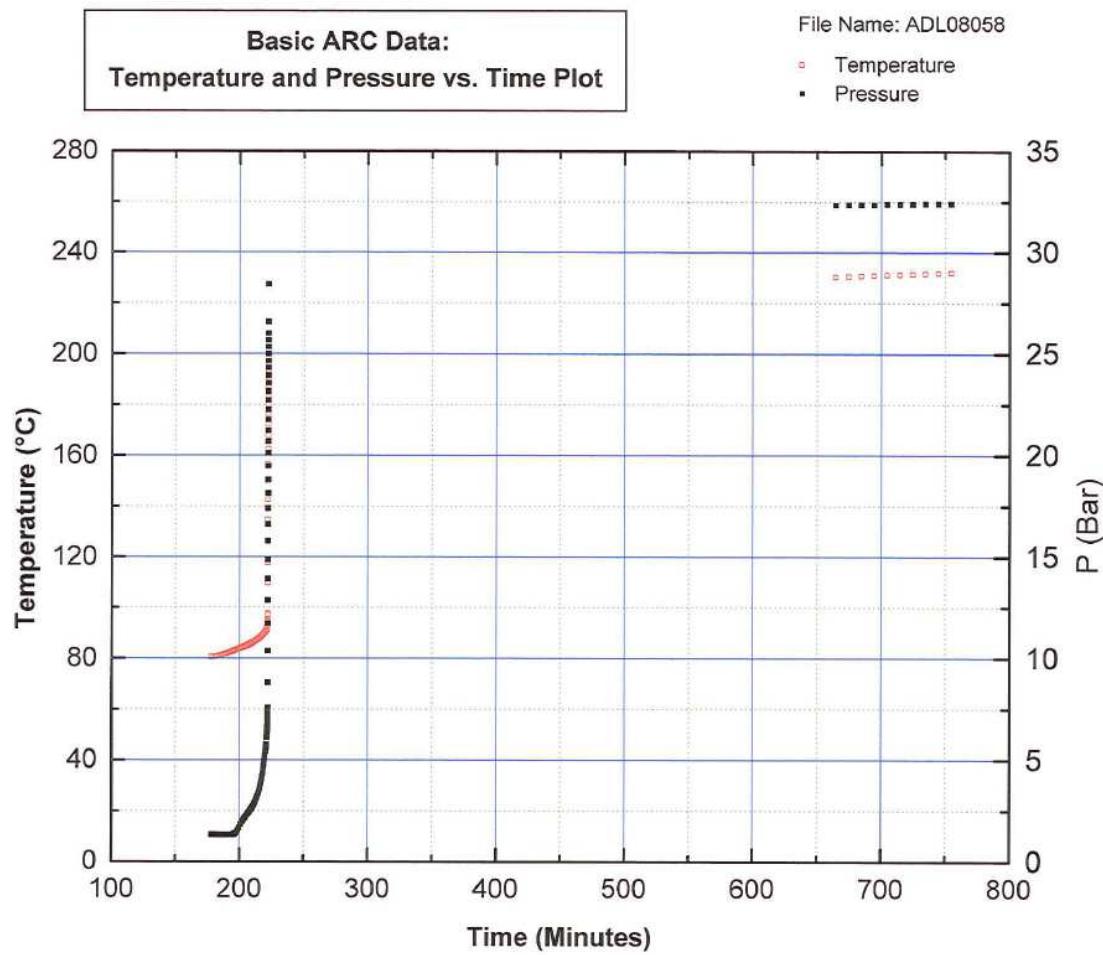
AIBN

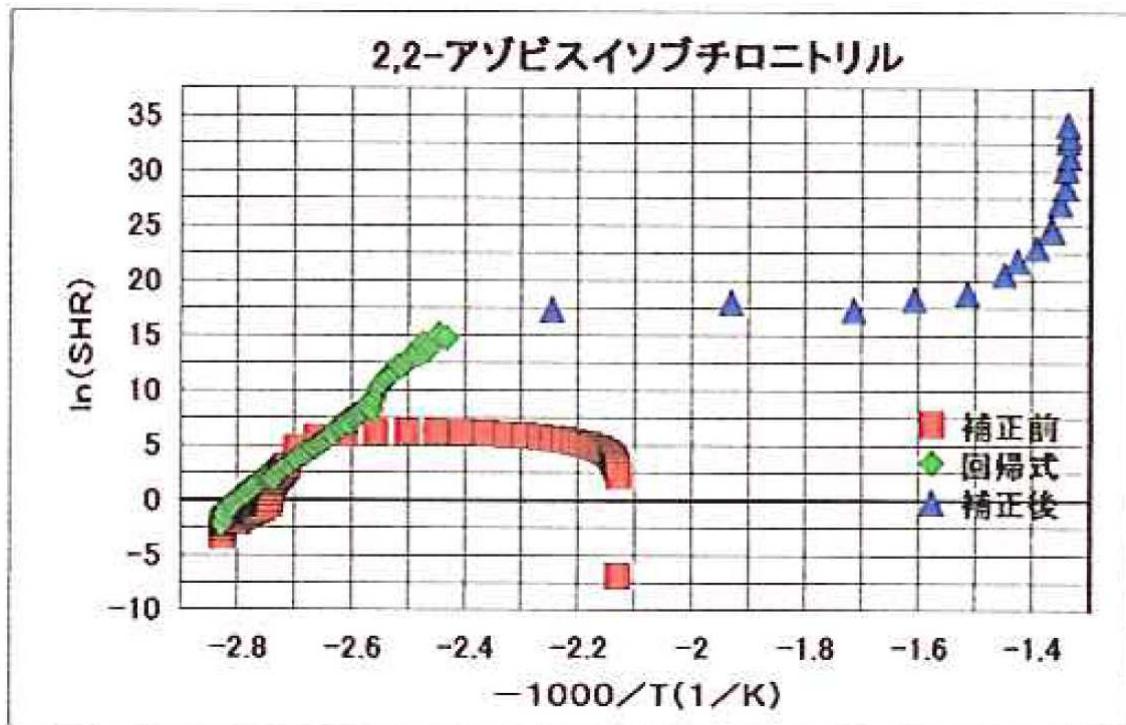


ARC device: ARC2000 (Arthur D. Little Inc.)

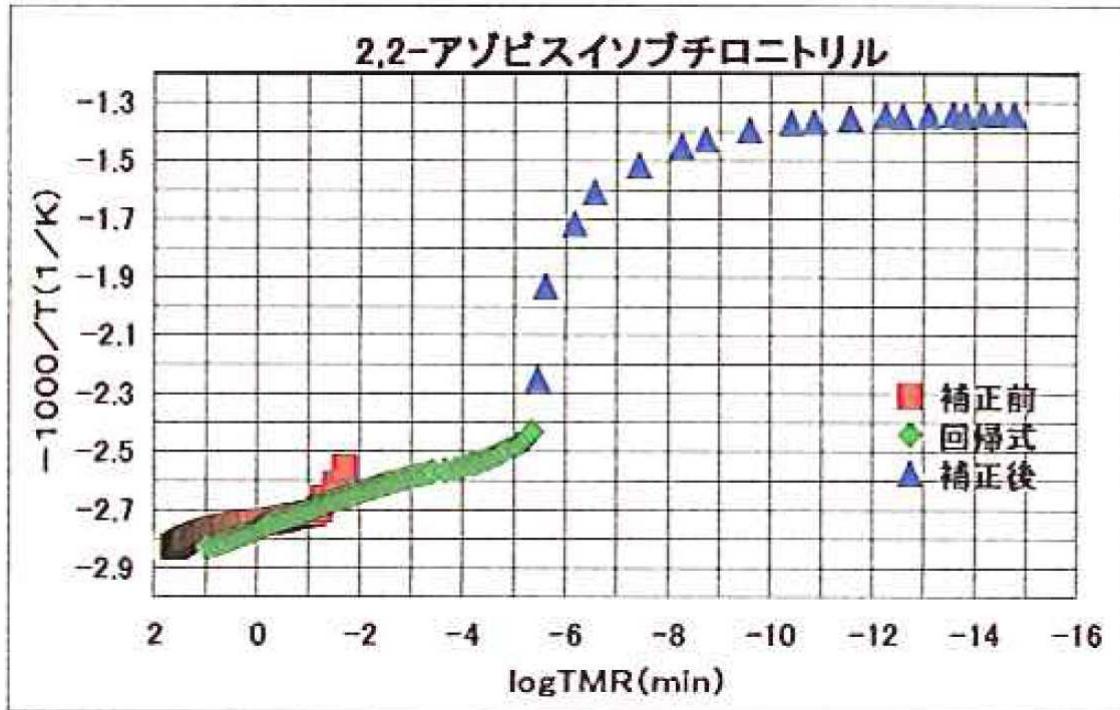
Date: 2008/12/17

Operator: KJ

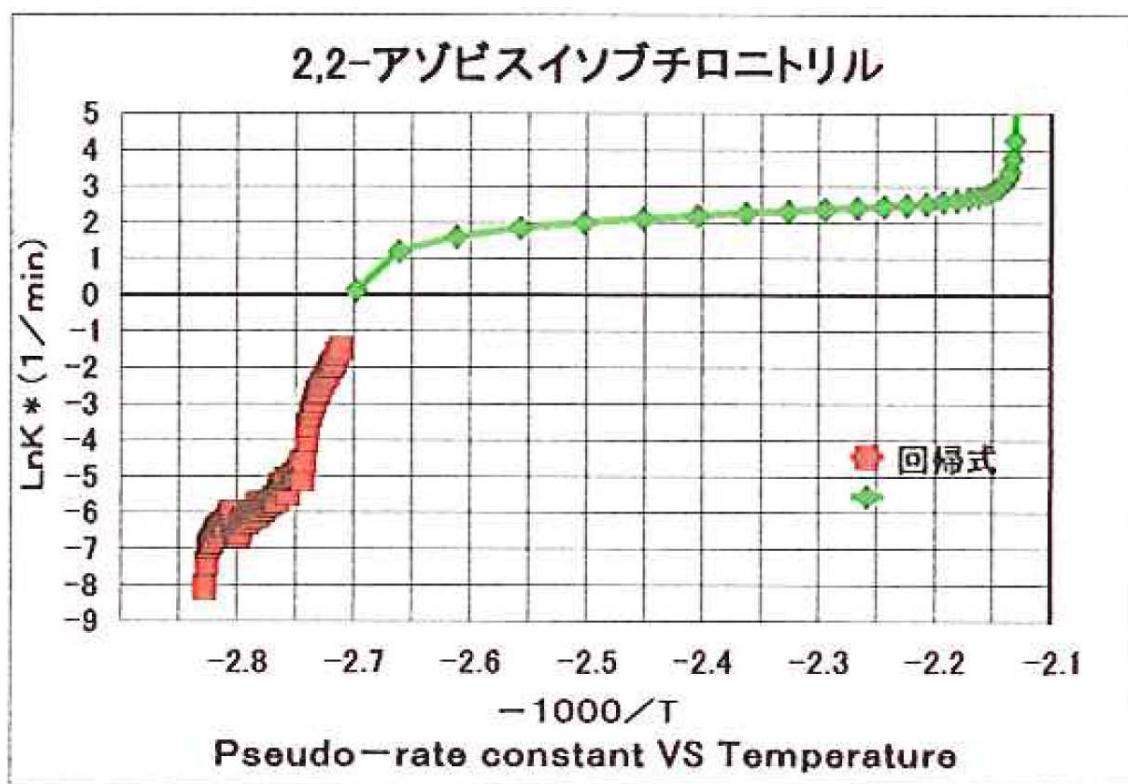




Temperature vs. Self heating rate



TMR vs. Temperature



Arrhenius equation (approximate calculation)

	Date	2008/12/17
Measuring conditions	ARC device	ARC2000 (Arthur D. Little Inc.)
	Operating Institute	KJ
	Operator	KJ
	Material of Bomb	Hastelloy C
	Weight of Bomb (g)	15.136
	Volume of Bomb (mL)	about 9
	Weight of sample (g)	1.301
	Weight of residue (g)	1.043
	Specific heat of Bomb ($J K^{-1} g^{-1}$)	0.419
	Specific heat of sample ($J K^{-1} g^{-1}$)	2.093
	ϕ facotr	3.33
	Start temperature (°C)	50
	End temperature (°C)	300
	Temperature increment (K)	5
	Waiting time (min)	10

	Searching time (min)	10
	Exothermic threshold (K min^{-1})	0.02
	Logging intervals ($^{\circ}\text{C}$)	0.2
	Pressure limit (kPa)	17000
	Atmosphere	Air, atmospheric pressure
Results	T_o , Exothermic temperature ($^{\circ}\text{C}$)	80.49
	Self heating rate at T_o (K min^{-1})	0.036
	Pressure at T_o (kPa)	130
	Temperature at maximum self heating rate ($^{\circ}\text{C}$)	126.39
	Maximum self heating rate (K min^{-1})	515.03
	Pressure at maximum self heating rate (kPa)	1390
	Pressure rising rate at maximum self heating rate (kPa min^{-1})	6308
	Maximum pressure (kPa)	2840
	Maximum pressure rising rate (kPa min^{-1})	9524
	Temperature at maximum pressure rising rate ($^{\circ}\text{C}$)	102.59
	Time to maximum rate (min)	43.94
	Maximum temperature ($^{\circ}\text{C}$)	196.65
	Adiabatic temperature rise ($^{\circ}\text{C}$)	116.16
	Activation energy (kJ mol^{-1})	364.7
	Heat of decomposition (J g^{-1})	808.1
Corrected results	T_{ARC} , Exothermic temperature ($^{\circ}\text{C}$)	73.63
	Time of maximum rate at T_{ARC} (min)	118.30
	Self heating rate at T_{ARC} (K min^{-1})	0.02
	Maximum self heating rate (K min^{-1})	7.37×10^{14}
	Maximum temperature ($^{\circ}\text{C}$)	474.82
	Adiabatic temperature rise ($^{\circ}\text{C}$)	401.19
	Heat of decomposition (J g^{-1})	841.5